# Synthesis, Nicotinic Acetylcholine Receptor Binding, and Antinociceptive Properties of 3'-(Substituted Phenyl)epibatidine Analogues. Nicotinic Partial Agonists ${ }^{\perp}$ 

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#### Abstract

In 1992, John Daly et al. reported the isolation and structure determination of epibatidine. Epibatidine's unique structure and its potent nicotinic agonist activity have had a tremendous impact on nicotine receptor research. This research has led to a better understanding of the nicotinic acetylcholine receptor ( nAChR ) pharmacophore and to epibatidine analogues with potential as pharmacotherapies for treating various CNS disorders. In this study, we report the synthesis, receptor binding ( $\left[{ }^{3} \mathrm{H}\right]$ epibatidine and $\left[{ }^{125} I\right]$ iodoMLA), and in vivo pharmacological properties (mouse tail flick, hot plate, hypothermia, and spontaneous activity) of a series of $3^{\prime}$-(substituted phenyl)epibatidine analogues ( $\mathbf{5 a} \mathbf{a} \mathbf{- m}$ ). Results from these studies have added to the understanding of the nAChR pharmacophore and led to nicotinic partial agonists that may have potential for smoking cessation. All the analogues had affinities for the $\alpha 4 \beta 2 \mathrm{nAChR}$ similar to epibatidine (1). $3^{\prime}$-(3-Dimethylaminophenyl)epibatidine ( $\mathbf{5 m}$ ) has a nicotinic partial agonist pharmacological profile similar to the smoking cessation drug varenicline. Other analogues are partial agonists with varying degrees of nicotinic functional agonist and antagonist activity. $3^{\prime}$-(3-Aminophenyl)epibatidine $(\mathbf{5 j})$ is a more potent functional agonist and antagonist in all tests than varenicline. $3^{\prime}$-(3-Fluorophenyl)epibatidine and $3^{\prime}$-(3-chlorophenyl)epibatidine ( $\mathbf{5 c}$ and 5e) are more potent than varenicline when tested as agonists in four pharmacological tests and antagonists when evaluated against nicotine in the analgesia hot-plate test.


Since it is estimated that four million smoking-related deaths result annually from smoking-related diseases such as lung cancer, chronic obstructive pulmonary disease (COPD), and cardiovascular disease, ${ }^{1}$ there is great interest in the development of pharmacotherapies for aiding people to stop smoking. ${ }^{2}$ In addition to nicotine (2) replacement therapy (NRT), the $\alpha 4 \beta 2 \mathrm{nAChR}$ partial agonist varenicline (3) and the antidepressant bupropion (4), which is also a noncompetitive nAChR antagonist, are the first-line treatment drugs for smoking cessation. ${ }^{2-4}$

Epibatidine (1), a structurally novel nicotinic acetylcholinergic compound, was isolated by Daly et al. from the skin of the Ecuadorian poison frog Epipedobates tricolor. ${ }^{5}$ Its unique structure and potent nicotinic acetylcholine receptor (nAChR) activity has had a major impact on nicotinic receptor research. A SciFinder search on epibatidine reveals 1013 references from 1992 (original report on the isolation by Daly et al.) to September 2009. Even though epibatidine's acute toxicity limited its therapeutic potential, ${ }^{6-9}$ it has served as a lead structure to develop pharmacotherapies for treating various CNS disorders including Alzheimer's and Parkinson's diseases, pain, schizophrenia, anxiety, depression, Tourette's syndrome, and smoking cessation. ${ }^{10}$

During the last several years, we have conducted structure-activity relationship (SAR) studies using epibatidine (1) as our lead structure to help characterize pharmacophores for the nAChR and to identify nAChR agonists, partial agonists, and antagonists as potential pharmacotherapies for treating smokers. ${ }^{11-19}$ In a preliminary study, we reported that introduction of a phenyl group at the $3^{\prime}$-position on the 2-chloropyridine ring of epibatidine gave $\mathbf{5 a}$, which had high affinity for $\alpha 4 \beta 2$ nAChR but was $100-350$-times less potent than epibatidine (1) in the mouse tail-flick, hot-plate, hypothermia, and spontaneous-activity tests after acute administration. ${ }^{11}$ The ability of 5a to antagonize nicotine-induced antinociception was not tested

[^0]in this study. In the present investigation, we report the nicotinic antagonist properties of $\mathbf{5 a}$ and compare the nAChR binding and agonist/antagonist pharmacological properties of the $3^{\prime}$-(substituted phenyl)epibatidine analogues $\mathbf{5 a}-\mathbf{m}$ to those of the nAChR agonists nicotine (2) and epibatidine (1) and the partial agonist varenicline (3). All analogues $(\mathbf{5 a} \mathbf{- m})$ had high affinity for the $\alpha 4 \beta 2$ nAChR similar to epibatidine (1). Also like epibatidine, they also had weak affinity for the $\alpha 7$ nAChR. Compounds $\mathbf{5 a} \mathbf{- m}$ showed both agonist and antagonist activity in the mouse acute tail-flick, hot-plate, hypothermia, and spontaneous-activity functional tests and, thus, are partial nAChR agonists.

(1R,2R,4S)-epibatidine (1)


5a, $X=Y=H$
b, $X=F, Y=H$
c, $X=H, Y=F$
d, $X=C I, Y=H$
e, $X=H, Y=C l$
f, $X=H, Y=B r$
g, $X=\mathrm{NO}_{2}, Y=\mathrm{H}$
h, $X=\mathrm{H}, \mathrm{Y}=\mathrm{NO}_{2}$
i, $X=\mathrm{NH}_{2}, Y=\mathrm{H}$
j, $\mathrm{X}=\mathrm{H}, \mathrm{Y}=\mathrm{NH}_{2}$
k, $\mathrm{X}=\mathrm{CH}_{3} \mathrm{O}, \mathrm{Y}=\mathrm{H}$
I, $X=\mathrm{H}, \mathrm{Y}=\mathrm{CH}_{3} \mathrm{O}$
m, $X=H, Y=\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~N}$

## Scheme $1^{a}$


${ }^{a}$ Reagents: (a) $\mathrm{Pd}(\mathrm{OAc})_{2}, \mathrm{P}(o \text {-tolyl })_{3}, \mathrm{Na}_{2} \mathrm{CO}_{3},(\mathrm{X}) \mathrm{C}_{6} \mathrm{H}_{4} \mathrm{~B}(\mathrm{OH})_{2}$, DME; (b) $\mathrm{CF}_{3} \mathrm{CO}_{2} \mathrm{H}$.

## Results and Discussion

Chemistry. The synthesis of $\mathbf{5 b}, \mathbf{5 d}, \mathbf{5 g}, \mathbf{5 i}$, and $\mathbf{5 k}$ is shown in Scheme 1. Palladium acetate-catalyzed coupling of tert-butoxycar-bonyl-3'-bromoepibatidine $(6)^{14}$ with the appropriately substituted phenylboronic acid in dimethoxyethane (DME) in the presence of tris( $o$-tolyl)phosphine and sodium carbonate gave the tert-butoxy-carbonyl-protected $3^{\prime}$-(substituted phenyl)epibatidine analogues (7). Treatment of $7 \mathbf{7 a}-\mathbf{e}\left(\mathrm{X}=\mathrm{F}, \mathrm{Cl}, \mathrm{NO}_{2}, t \mathrm{BocNH}\right.$, and $\left.\mathrm{CH}_{3} \mathrm{O}\right)$ with trifluoroacetic acid in methylene chloride removed the protecting tert-butoxycarbonyl group and afforded the desired $3^{\prime}$-(substituted phenyl)epibatidine analogues $\mathbf{5 b}, \mathbf{5 d}, \mathbf{5 g}, \mathbf{5 i}$, and $\mathbf{5 k}$.
Scheme 2 outlines the synthesis of $\mathbf{5 j}$ and $\mathbf{5 m}$ starting with the previously reported $3^{\prime}$-(3-nitrophenyl)epibatidine (5h). ${ }^{19}$ Reduction of $\mathbf{5 h}$ with iron powder in ethanolic hydrogen chloride gave the desired $\mathbf{5} \mathbf{j}$. In order to prepare $\mathbf{5 m}, \mathbf{5 h}$ was first converted to the tert-butoxycarbonyl-protected $\mathbf{8}$ using tert-butoxycarbonyl anhydride catalyzed by dimethylaminopyridine (DMAP) in methylene chloride containing a small amount of triethylamine. Reduction of 8 with nickel borohydride and hydrochloric acid in methanol provided the amino compound 9 , which was reductively methylated to the 7-tert-butoxycarbonyl-protected dimethylamino compound 10 using sodium cyanoborohydride and formaldehyde in acetonitrile. Treatment of $\mathbf{1 0}$ with trifluoroacetic acid in methylene chloride afforded $\mathbf{5 m}$. Target compounds $\mathbf{5 c}, \mathbf{5 e}$, and $\mathbf{5 f}$ were all prepared from $\mathbf{5 j}$ using various diazotization procedures (Scheme 3). Thus, diazotization of $\mathbf{5 j}$ using sodium nitrite in $70 \%$ hydrogen fluoridepyridine yielded $\mathbf{5 c}$, and diazotization of $\mathbf{5 j}$ using $n$-butyl nitrite with cuprous chloride or cuprous bromide in acetonitrile afforded 5e and 5f, respectively.
Biological Activity. The nAChR binding affinities and the functional nicotinic pharmacological properties of several $3^{\prime}$ (substituted phenyl)epibatidine analogues were determined. The $K_{\mathrm{i}}$ values for the inhibition of $\left[{ }^{3} \mathrm{H}\right]$ epibatidine and $\left[{ }^{[25} \mathrm{I}\right]$ iodoMLA binding at the $\alpha 4 \beta 2$ and $\alpha 7 \mathrm{nAChRs}$, respectively, for compounds $\mathbf{5 a}-\mathbf{m}$ along with reference compounds ( + )- and ( - )-epibatidine $[(+)-\mathbf{1}$ and ( - )-1], nicotine (2), and varenicline (3) are listed in Table 1. (+)- and (-)-Epibatidine with $K_{\mathrm{i}}$ values of 0.026 and 0.018 nM , respectively, have very similar affinities for the $\alpha 4 \beta 2 \mathrm{nAChR} .{ }^{15}$ Nicotine and varenicline have $K_{\mathrm{i}}$ values of 1.5 and 0.12 nM ,

Scheme $\mathbf{2}^{a}$

${ }^{a}$ Reagents: (a) $\mathrm{Fe}, \mathrm{HCl}, \mathrm{C}_{2} \mathrm{H}_{5} \mathrm{OH}$; (b) $(\mathrm{Boc})_{2} \mathrm{O}$, DMAP, $\left(\mathrm{C}_{2} \mathrm{H}_{5}\right)_{3} \mathrm{~N}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$; (c) $\mathrm{Ni}_{2} \mathrm{~B}, \mathrm{CH}_{3} \mathrm{OH}, \mathrm{HCl}$; (d) $\mathrm{NaCNBH}_{3}, \mathrm{H}_{2} \mathrm{CO}, \mathrm{CH}_{3} \mathrm{CN}$; (e) $\mathrm{CF}_{3} \mathrm{CO}_{2} \mathrm{H}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$.

## Scheme $3^{a}$


${ }^{a}$ Reagents: (a) HF pyridine, $\mathrm{NaNO}_{2}$; (b) $\mathrm{nC}_{4} \mathrm{H}_{9} \mathrm{ONO}, \mathrm{CuCl}, \mathrm{CH}_{3} \mathrm{CN}, 65^{\circ} \mathrm{C}$; (c) $\mathrm{nC}_{4} \mathrm{H}_{9} \mathrm{ONO}, \mathrm{CuBr}, \mathrm{CH}_{3}, \mathrm{CN}, 65^{\circ} \mathrm{C}$.
respectively. Since the affinities of the epibatidine isomers are so similar, the (substituted phenyl)epibatidine analogues $\mathbf{5 a}-\mathbf{m}$ are compared only to the natural epibatidine isomer. The unsubstituted phenyl analogue 5 a has a $K_{\mathrm{i}}$ of 0.021 nM at the $\alpha 4 \beta 2 \mathrm{nAChR}$, which is almost identical to that of epibatidine ( $K_{\mathrm{i}}=0.026 \mathrm{nM}$ ), and has 71- and 6 -times higher affinity at the $\alpha 4 \beta 2 \mathrm{nAChR}$ than nicotine and varenicline, respectively. Substitution of the $3^{\prime}$-phenyl ring of $5 \mathbf{5}$ with a 4 - or 3-position electron-withdrawing or -releasing substituent had only small effects on binding affinity at the $\alpha 4 \beta 2$ nAChR . The $K_{\mathrm{i}}$ values varied from 0.008 and 0.009 nM for the 3-nitrophenyl ( $\mathbf{5 h}$ ) and 3-dimethylamino ( $\mathbf{5 m}$ ) analogues to 0.034 and 0.039 nM for the 4 -aminophenyl ( $\mathbf{5 i}$ ) and the 4-chlorophenyl (5d) analogues. In every case the $3^{\prime}$-(3-substituted phenyl) analogue had a slightly lower $K_{\mathrm{i}}$ value than the corresponding $3^{\prime}$-(4-substituted phenyl) analogue (compare $\mathbf{5 c}, \mathbf{5 e}, \mathbf{5 h}, \mathbf{5 j}$, and $\mathbf{5 1}$ to the corresponding $\mathbf{5 b}, \mathbf{5 d}, \mathbf{5 g}, \mathbf{5 i}$, and $\mathbf{5 k}$ ). Similar to epibatidine, all analogues had relatively weak affinity for the $\alpha 7 \mathrm{nAChR}$. The $\alpha 7$ $\mathrm{nAChR} K_{\mathrm{i}}$ values varied from 30.5 nM for $\mathbf{5 e}$ to 1100 nM for $\mathbf{5 m}$, compared to a $K_{\mathrm{i}}$ of 198 nM for epibatidine and 32.5 nM for varenicline. The 4-chlorophenyl analogue $\mathbf{5 e}$, with a $K_{\mathrm{i}}$ of 30.5 nM , had the highest affinity for the $\alpha 7 \mathrm{nAChR}$ but was still greater than 2000 -fold selective for $\alpha 4 \beta 2 \mathrm{nAChR}$ relative to $\alpha 7 \mathrm{nAChR}$.

Natural epibatidine has an $\mathrm{ED}_{50}$ of 0.0061 and $0.004 \mathrm{mg} / \mathrm{kg}$ in the tail-flick and hot-plate antinociception tests. ${ }^{15}$ The unnatural epibatidine isomer has an $\mathrm{ED}_{50}$ of 0.0066 in the tail-flick test. ${ }^{15}$ Epibatidine also has $\mathrm{ED}_{50}$ values of 0.004 and $0.001 \mathrm{mg} / \mathrm{kg}$ in the

Table 1. Comparison of Nicotine, Epibatidine, and Varenicline Radioligand Binding and Antinociception Data to $3^{\prime}$-(Substituted Phenyl)epibatidine Analogues (5a-m)

|  |  | , | ( |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  | $\mathrm{ED}_{50}$ |  | $\mathrm{ED}_{50}$ | $\mathrm{ED}_{50}$ <br> $\mathrm{mg} / \mathrm{kg}$ | $\mathrm{AD}_{50}$ | $(\mu \mathrm{g} / \mathrm{kg})^{c}$ |
| compound | X | Y | $\left[{ }^{3} \mathrm{H}\right]$ epibatidine $\left(K_{\mathrm{i}}, \mathrm{nM}\right)$ | $\begin{gathered} {\left[\begin{array}{c} \left.{ }^{125} \Gamma\right] \text { iodoMLA } \\ \left(K_{\mathrm{i}}, \mathrm{nM}\right)^{b} \end{array}\right.} \end{gathered}$ | $\underset{\text { tail-flick }^{\mathrm{mg}} / \mathrm{kg}}{ }$ | $\mathrm{mg} / \mathrm{kg}$ <br> hot-plate ${ }^{c}$ | $\mathrm{mg} / \mathrm{kg}$ hypothermia ${ }^{c}$ | spontaneous activity ${ }^{\text {c }}$ | tail-flick | hot-plate |
| $\begin{gathered} \text { (+)-epibatidine } \\ {[(+)-1]^{d}} \end{gathered}$ |  |  | $0.026 \pm 0.002^{e}$ |  | 0.0061 | 0.004 | 0.004 | 0.001 |  |  |
| $\begin{gathered} (-) \text {-epibatidine } \\ {[(-)-1]^{d}} \end{gathered}$ |  |  | $0.018 \pm 0.001^{e}$ |  | $0.0066{ }^{e}$ |  |  |  |  |  |
| Nicotine (2) ${ }^{e}$ |  |  | $1.5 \pm 0.3$ |  | 1.3 | 0.65 | 1.0 | 0.5 |  |  |
| varenicline (3) ${ }^{f}$ |  |  | $0.12 \pm 0.02$ | $32.5 \pm 1.3$ | 11\% @ 10 | 10\% @ 10 | 2.8 | 2.1 | 0.2 | 470 |
| 5a | H | H | $0.021 \pm 0.005$ | $260 \pm 5.0$ | $\begin{aligned} & 0.7^{g} \\ & (0.5-1.0) \end{aligned}$ | $\begin{aligned} & 1.0^{g} \\ & (0.5-2.0) \end{aligned}$ | $\begin{aligned} & 0.4^{g} \\ & (0.1-0.9) \end{aligned}$ | $\begin{aligned} & 0.35^{g} \\ & (0.2-0.85) \end{aligned}$ | $\begin{aligned} & 280 \\ & (80-900) \end{aligned}$ | 30\% @ 10,000 |
| 5b | F | H | $0.017 \pm 0.003$ | $309 \pm 13$ | $\begin{aligned} & 2.48 \\ & (1.7-3.5) \end{aligned}$ | $\begin{aligned} & 1.27 \\ & (0.8-1.9) \end{aligned}$ | $\begin{aligned} & 0.33 \\ & (0.22-0.67) \end{aligned}$ | $\begin{aligned} & 0.33 \\ & (0.15-0.7) \end{aligned}$ | $\begin{aligned} & 0.5 \\ & (0.06-29) \end{aligned}$ | 30\% @ 100 |
| 5c | H | F | $0.012 \pm 0.001$ | $250 \pm 44$ | $\begin{aligned} & 1.0 \\ & (0.7-1.5) \end{aligned}$ | $\begin{aligned} & 0.43 \\ & (0.2-0.94) \end{aligned}$ | $\begin{aligned} & 0.19 \\ & (0.1-0.7) \end{aligned}$ | $\begin{aligned} & 0.07 \\ & (0.05-0.10) \end{aligned}$ | $\begin{aligned} & 3.9 \\ & (5-30) \end{aligned}$ | $\begin{aligned} & 86 \\ & (20-300) \end{aligned}$ |
| 5d | Cl | H | $0.039 \pm 0.005$ | $209 \pm 50$ | $\begin{aligned} & 2.6 \\ & (2.3-3.3) \end{aligned}$ | $\begin{aligned} & 2 \\ & (1.4-2.7) \end{aligned}$ | $\begin{aligned} & 0.89 \\ & (0.5-1.2) \end{aligned}$ | $\begin{aligned} & 0.53 \\ & (0.3-0.9) \end{aligned}$ | $\begin{aligned} & 1 \\ & (0.1-27) \end{aligned}$ | 10\% @ 100 |
| 5e | H | Cl | $0.013 \pm 0.001$ | $30.5 \pm 84$ | $\begin{aligned} & 3.4 \\ & (2.6-4.3) \end{aligned}$ | $\begin{aligned} & 4.2 \\ & (1.5-11.6) \end{aligned}$ | $\begin{aligned} & 0.38 \\ & (0.2-1) \end{aligned}$ | $\begin{aligned} & 0.06 \\ & (0.02-0.13) \end{aligned}$ | $\begin{aligned} & 2.2 \\ & (4-15) \end{aligned}$ | $\begin{aligned} & 80 \\ & (30-200) \end{aligned}$ |
| 5f | H | Br | $0.016 \pm 0.003$ | $85 \pm 10$ | $\begin{aligned} & 0.54 \\ & (0.44-0.67) \end{aligned}$ | $\begin{aligned} & 2.2 \\ & (1.2-3.8) \end{aligned}$ | $\begin{aligned} & 0.67 \\ & (0.5-0.9) \end{aligned}$ | $\begin{aligned} & 0.11 \\ & (0.06-0.22) \end{aligned}$ | $\begin{aligned} & 10 \\ & (7-90) \end{aligned}$ | $\begin{aligned} & 220 \\ & (50-600) \end{aligned}$ |
| 5 g | $\mathrm{NO}_{2}$ | H | $0.015 \pm 0.001$ | $250 \pm 25$ | $\begin{aligned} & 0.46 \\ & (0.33-0.64) \end{aligned}$ | $\begin{aligned} & 0.38 \\ & (0.2-0.5) \end{aligned}$ | $\begin{aligned} & 0.23 \\ & (0.1-0.6) \end{aligned}$ | $\begin{aligned} & 0.24 \\ & (0.1-0.5) \end{aligned}$ | $\begin{aligned} & 7 \\ & (3-10) \end{aligned}$ | 17\% @ 50 |
| 5h | H | $\mathrm{NO}_{2}$ | $0.008 \pm 0.0003$ | $229 \pm 43$ | $\begin{aligned} & 4.6 \\ & (3.3-6.5) \end{aligned}$ | $\begin{aligned} & 3.2 \\ & (2.4-4.5) \end{aligned}$ | $\begin{aligned} & 1.4 \\ & (1-2.4) \end{aligned}$ | $\begin{aligned} & 0.9 \\ & (0.7-1.2) \end{aligned}$ | $\begin{aligned} & 0.25 \\ & (0.02-0.7) \end{aligned}$ | $\begin{aligned} & 180 \\ & (29-1078) \end{aligned}$ |
| 5 i | $\mathrm{NH}_{2}$ | H | $0.034 \pm 0.001$ | $234 \pm 17$ | $\begin{aligned} & 5 \\ & (3.4-7.2) \end{aligned}$ | $\begin{aligned} & 4.5 \\ & (3.6-5.7) \end{aligned}$ | $\begin{aligned} & 1.1 \\ & (0.8-1.9) \end{aligned}$ | $\begin{aligned} & 0.52 \\ & (0.11-2.3) \end{aligned}$ | $\begin{aligned} & 6 \\ & (5-7) \end{aligned}$ | 0\% @ 100 |
| 5j | H | $\mathrm{NH}_{2}$ | $0.017 \pm 0.001$ | $256 \pm 46$ | $\begin{aligned} & 0.34 \\ & (0.24-0.46) \end{aligned}$ | $\begin{aligned} & 0.3 \\ & (0.17-0.53) \end{aligned}$ | $\begin{aligned} & 0.31 \\ & (0.2-0.5) \end{aligned}$ | $\begin{aligned} & 0.03 \\ & (0.009-0.12) \end{aligned}$ | $\begin{aligned} & 0.14 \\ & (0.05-0.4) \end{aligned}$ | $\begin{aligned} & 26 \\ & (5-90) \end{aligned}$ |
| 5k | $\mathrm{CH}_{3} \mathrm{O}$ | H | $0.022 \pm 0.008$ | $175 \pm 10$ | $\begin{aligned} & 2.36 \\ & (2-3.7) \end{aligned}$ | $\begin{aligned} & 1.1 \\ & (0.5-2.7) \end{aligned}$ | $\begin{aligned} & 0.79 \\ & (0.55-1.1) \end{aligned}$ | $\begin{aligned} & 0.43 \\ & (0.1-1.3) \end{aligned}$ | $\begin{aligned} & 10 \\ & (1-9) \end{aligned}$ | 10\%@200 |
| 51 | H | $\mathrm{CH}_{3} \mathrm{O}$ | $0.019 \pm 0.005$ | $558 \pm 50$ | $\begin{aligned} & 0.7 \\ & (0.3-1.2) \end{aligned}$ | $\begin{aligned} & 0.8 \\ & (0.3-1.7) \end{aligned}$ | $\begin{aligned} & 0.7 \\ & (0.5-1.1) \end{aligned}$ | $\begin{aligned} & 0.2 \\ & (0.18-0.23) \end{aligned}$ | $\begin{aligned} & 29 \\ & (10-60) \end{aligned}$ | 0\% @ 100 |
| 5m | H | $\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~N}$ | $0.009 \pm 0.001$ | $1100 \pm 157$ | 0\% @ 2 | 0\%@2 | $\begin{aligned} & 3.3 \\ & (2.1-5.6) \end{aligned}$ | $\begin{aligned} & 2.0 \\ & (0.7-5.8) \end{aligned}$ | $\begin{aligned} & 20 \\ & (10-50) \end{aligned}$ | $\begin{aligned} & 560 \\ & (100-9200) \end{aligned}$ |

${ }^{a}$ All epibatidine analogues were tested as hydrochloride salts, and all were racemates. ${ }^{b}$ Data represent means $\pm$ SE from at least three independent experiments. ${ }^{c}$ Results are provided as $\mathrm{ED}_{50}$ or $\mathrm{AD}_{50}$ values ( $\pm$ confidence limits) or as a percent effect at the individual dose. ${ }^{d}$ Compound ( + )- $\mathbf{1}$ is the natural epibatidine hydrochloride, and $(-)-\mathbf{1}$ is the enantiomeric epibatidine hydrochloride. ${ }^{e}$ Data taken from ref 15 . ${ }^{f}$ Data taken from ref 19 . ${ }^{g}$ Data taken from ref 13.
hypothermia and spontaneous-activity test. The $\mathrm{ED}_{50}$ values for nicotine (2) in the tail-flick, hot-plate, hypothermia, and spontane-ous-activity tests are $1.3,0.65,1.0$, and $0.5 \mathrm{mg} / \mathrm{kg}$, respectively. Varenicline (3) is inactive in the tail-flick and hot-plate tests but has $\mathrm{ED}_{50}$ values of 2.8 and $2.1 \mathrm{mg} / \mathrm{kg}$ in the hypothermia and spontaneous-activity tests. In addition, varenicline (3) antagonizes the nicotine-induced antinociception in the tail-flick and hot-plate tests with $\mathrm{AD}_{50}$ values of 0.2 and $470 \mu \mathrm{~g} / \mathrm{kg}$, respectively. ${ }^{19}$
With the exception of the 3-dimethylaminophenyl analogue 5 m , which had a profile like varenicline, all compounds showed weak agonist activity compared to epibatidine in the tail-flick, hot-plate, hypothermia, and spontaneous-activity tests. The 3 -aminophenyl analogue $\mathbf{5 j}$, with $\mathrm{ED}_{50}$ values of $0.34,0.3,0.31$, and $0.03 \mathrm{mg} / \mathrm{kg}$ in the tail-flick, hot-plate, and hypothermia tests, was the most potent analogue; however, this potency is $57-$, $75-$, 76 - and $30-$ times weaker than epibatidine (1) in these four tests. The agonist potencies of these analogues were more similar to nicotine (2) than epibatidine. Similar to varenicline (3), $\mathbf{5 m}$ did not have agonist activity in the tail-flick and hot-plate tests but had $\mathrm{ED}_{50}$ values of 3.3 and $2.0 \mathrm{mg} / \mathrm{kg}$ in the hypothermia and spontaneous-activity tests, respectively, compared to $\mathrm{ED}_{50}$ values of 2.8 and $2.1 \mathrm{mg} / \mathrm{kg}$ for varenicline (3) in these two tests. Most analogues did not exhibit pharmacological selectivity. For the most part they all produced similar potencies in all four tests. The most variation was in the spontaneous-activity test. For example, the 3-aminophenyl analogue
$\mathbf{5 j}$ was approximately an order of magnitude more potent in this test than in the other three tests.

The high binding affinity of compounds $\mathbf{5 a}-\mathbf{m}$ for $\alpha 4 \beta 2$ nAChRs combined with the relatively weak agonist potency suggested that these analogues might act as nAChR functional antagonists in vivo. Indeed, all analogues were potent antagonists with $\mathrm{AD}_{50}$ values of $0.14-20 \mu \mathrm{~g} / \mathrm{kg}$ in the tail-flick test compared to $0.2 \mu \mathrm{~g} / \mathrm{kg}$ for the partial agonist varenicline (3). In addition, the $3^{\prime}$-phenyl and $3^{\prime}$ (substituted phenyl) analogues $\mathbf{5 a}, \mathbf{5 c}, \mathbf{5 e}, \mathbf{5} \mathbf{h}, \mathbf{5 j}$, and $\mathbf{5 m}$, respectively, were also potent antagonists in the hot-plate test with $\mathrm{AD}_{50}$ values of $26-560 \mu \mathrm{~g} / \mathrm{kg}$ compared to $470 \mu \mathrm{~g} / \mathrm{kg}$ for varenicline (3). The three most potent analogues were the 3 -aminophenyl (5j), 3-chlorophenyl (5e), and 3-fluorophenyl (5c), with $\mathrm{AD}_{50}$ values of 26,80 , and $86 \mu \mathrm{~g} / \mathrm{kg}$, respectively, which is $18-$, $5.9-$, and 5.5 -times more potent as an antagonist than varenicline in the hot-plate test.

In summary, the addition of a $3^{\prime}$-phenyl or electron-withdrawing or -releasing $3^{\prime}$-(3- or 4 -substituted phenyl) group to the highly potent nAChR agonist epibatidine (1) provided a series of analogues, $\mathbf{5 a - m}$. Like epibatidine these compounds had high affinity for the $\alpha 4 \beta 2 \mathrm{nAChR}$ and weak affinity for the $\alpha 7 \mathrm{nAChR}$. In contrast to the high potency of epibatidine (1), these analogues had agonist potency in the tail-flick, hot-plate, hypothermia, and spontaneous-activity tests in mice more like that of nicotine (2). On the other hand, like varenicline, these analogues were potent
antagonists in the tail-flick test and to a lesser degree in the hotplate test. Thus, similar to varenicline, these analogues are functional nAChR partial agonists in vivo. The 3-dimethylaminophenyl analogue $\mathbf{5 m}$ has an agonist/antagonist profile most like varenicline. The 3 -aminophenyl analogue $\mathbf{5} \mathbf{j}$ is a more potent functional agonist and antagonist in all tests than varenicline. The 3-fluorophenyl and 3-chlorophenyl analogues $\mathbf{5 c}$ and $\mathbf{5 e}$, respectively, are more potent than varenicline in all four agonist tests and the antagonist hotplate test. Thus, compounds $\mathbf{5 c}, \mathbf{5 e}, \mathbf{5 j}$, and $\mathbf{5 m}$ represent exciting lead structures for developing a new structural class of nicotinic partial agonists useful in the treatment of nicotine addiction (smokers) and possibly other CNS diseases and disorders.

## Experimental Section

General Experimental Procedures. Melting points were determined on a Mel-temp (Laboratory Devices, Inc.) capillary tube apparatus. NMR spectra were recorded on a Bruker Avance 300 using tetramethylsilane as internal standard. Thin-layer chromatography was carried out on Whatman silica gel 60 plates. Visualization was accomplished under UV or in an iodine chamber. Microanalysis was carried out by Atlantic Microlab, Inc. Flash chromatography was carried out using silica gel 60 ( $230-400$ mesh) using various solvent mixtures. CMA is $80 \%$ chloroform, $18 \%$ methanol, and $2 \%$ concentrated ammonium hydroxide.
The $\left[{ }^{3} \mathrm{H}\right]$ epibatidine was purchased from Perkin-Elmer Inc. (Boston, MA). The [ $\left.{ }^{125}\right]$ iodo-MLA was synthesized as previously reported. ${ }^{20}$

7-tert-Butoxycarbonyl-2-exo-[3-(4-fluorophenyl)-5-pyridinyl]-7azabicyclo[2.2.1]heptane (7a). To a resealable reaction tube were added compound $\mathbf{6}$ ( $231 \mathrm{mg}, 0.6 \mathrm{mmol}$ ), 4-fluorophenylboronic acid ( 168 mg , $1.2 \mathrm{mmol}), \mathrm{Pd}(\mathrm{OAc})_{2}(14 \mathrm{mg}, 0.06 \mathrm{mmol})$, tris $(o$-tolyl)phosphine ( 37 $\mathrm{mg}, 0.12 \mathrm{mmol})$, and $\mathrm{Na}_{2} \mathrm{CO}_{3}(159 \mathrm{mg}, 1.5 \mathrm{mmol})$ in DME ( 2 mL ) and $\mathrm{H}_{2} \mathrm{O}(0.5 \mathrm{~mL})$. The mixture was purged with argon, sealed, and heated in an $85^{\circ} \mathrm{C}$ oil bath overnight. The reaction mixture was cooled, filtered through Celite, and diluted with EtOAc. The organic phase was washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated. Flash chromatography of the resulting residue on a silica gel column using EtOAc-hexanes ( $1: 1$ ) yielded $217 \mathrm{mg}(90 \%)$ of 7 a as a yellow oil: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.40(9 \mathrm{H}, \mathrm{s}), 1.5-1.9(5 \mathrm{H}, \mathrm{m}), 2.02(1 \mathrm{H}, \mathrm{m}), 2.90$ $(1 \mathrm{H}, \mathrm{dd}, J=4.5,8.7 \mathrm{~Hz}), 4.23(1 \mathrm{H}, \mathrm{brs}), 4.39(1 \mathrm{H}, \mathrm{brs}), 7.1-7.2(2 \mathrm{H}$, $\mathrm{m}), 7.4-7.5(2 \mathrm{H}, \mathrm{m}), 7.64(1 \mathrm{H}, \mathrm{m}), 8.27(1 \mathrm{H}, \mathrm{m}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ $\delta 28.6,29.2,30.1,40.8,45.2,56.2,62.2,80.2,115.6\left(\mathrm{~d}, J_{\mathrm{CF}}=21.0\right.$ $\mathrm{Hz}), 131.5\left(\mathrm{~d}, J_{\mathrm{CF}}=8.7 \mathrm{~Hz}\right), 135.9,138.5,140.9,147.8,155.2,163.0$ (d, $J_{\mathrm{CF}}=247.7 \mathrm{~Hz}$ ).
$\mathbf{3}^{\prime}$-(4-Fluorophenyl)epibatidine (5b) Hydrochloride. Compound 7a ( $217 \mathrm{mg}, 0.54 \mathrm{mmol}$ ) was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{~mL})$. TFA ( 3 mL ) was added dropwise at $0{ }^{\circ} \mathrm{C}$ over 30 min . The mixture was stirred at room temperature for 4 h , poured into a cold solution of concentrated $\mathrm{NH}_{4} \mathrm{OH}-\mathrm{H}_{2} \mathrm{O}$ (1:1), and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic phase was washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and evaporated to dryness. Flash chromatography of the resulting residue on a silica gel column using $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$ yielded $136 \mathrm{mg}(83 \%)$ of $\mathbf{5 b}$ : ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta$ $1.5-1.7(5 \mathrm{H}, \mathrm{m}), 1.94(1 \mathrm{H}, \mathrm{dd}, J=9.0,12.0 \mathrm{~Hz}), 2.82(1 \mathrm{H}, \mathrm{dd}, J=$ $5.1,9.0 \mathrm{~Hz}), 3.62(1 \mathrm{H}, \mathrm{brs}), 3.81(1 \mathrm{H}$, brs $), 7.1-7.2(2 \mathrm{H}, \mathrm{m}), 7.4-7.5$ $(2 \mathrm{H}, \mathrm{m}), 7.76(1 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz}), 8.29(1 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 30.2,31.7,40.6,44.8,56.8,63.1,115.6\left(\mathrm{~d}, J_{\mathrm{CF}}=21.1 \mathrm{~Hz}\right)$, $131.6\left(\mathrm{~d}, J_{\mathrm{CF}}=8.7 \mathrm{~Hz}\right), 135.7,138.9,141.8,147.5,148.0,163.0(\mathrm{~d}$, $J_{\mathrm{CF}}=247.8 \mathrm{~Hz}$ ).

Compound 5b ( $136 \mathrm{mg}, 0.45 \mathrm{mmol}$ ) was dissolved in MeOH ( 4.6 $\mathrm{mL})$ at room temperature. $\mathrm{HCl}(1 \mathrm{M}$ in ether, 4.6 mL$)$ was added. After stirring for 30 min , the solvent was removed, and the residue was recrystallized from a MeOH -ether mixture to give $\mathbf{5 b} \cdot \mathrm{HCl}$ as a yellow solid: $\mathrm{mp}>200^{\circ} \mathrm{C}$ (dec); anal. C $60.06 \%$, H $5.16 \%$, N $8.07 \%$, calcd for $\mathrm{C}_{17} \mathrm{H}_{17} \mathrm{Cl}_{2} \mathrm{FN}_{2}$, C $60.19 \%$, $\mathrm{H} 5.05 \%$, N $8.26 \%$.

7-tert-Butoxycarbonyl-2-exo-[3-(4-chlorophenyl)-5-pyridinyl]-7azabicyclo[2.2.1]heptane (7b). To a resealable reaction tube were added compound 6 ( $233 \mathrm{mg}, 0.6 \mathrm{mmol}$ ), 4-chlorophenylboronic acid (188 $\mathrm{mg}, 1.2 \mathrm{mmol}), \mathrm{Pd}(\mathrm{OAc})(14 \mathrm{mg}, 0.06 \mathrm{mmol})$, tris $(o$-tolyl) phosphine ( $37 \mathrm{mg}, 0.12 \mathrm{mmol}$ ), and $\mathrm{Na}_{2} \mathrm{CO}_{3}(159 \mathrm{mg}, 1.5 \mathrm{mmol})$ in DME ( 2 $\mathrm{mL})$ and $\mathrm{H}_{2} \mathrm{O}(0.5 \mathrm{~mL})$. The mixture was purged with argon, sealed, and heated in an $85^{\circ} \mathrm{C}$ oil bath overnight. The reaction mixture was cooled, filtered through Celite, and diluted with EtOAc. The organic phase was washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated. Flash chromatography of the resulting residue on a silica gel column using

EtOAc-hexanes (1:1) yielded $247 \mathrm{mg}(98 \%)$ of $\mathbf{7 b}$ as a yellow oil: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.40(9 \mathrm{H}, \mathrm{s}), 1.5-1.9(5 \mathrm{H}, \mathrm{m}), 2.02(1 \mathrm{H}, \mathrm{dd}, J=9.0$, $12.3 \mathrm{~Hz}), 2.92(1 \mathrm{H}, \mathrm{dd}, J=4.8,9.0 \mathrm{~Hz}), 4.22(1 \mathrm{H}$, brs $), 4.38(1 \mathrm{H}$, brs), $7.3-7.5(4 \mathrm{H}, \mathrm{m}), 7.63(1 \mathrm{H}, \mathrm{d}, J=2.7 \mathrm{~Hz}), 8.28(1 \mathrm{H}, \mathrm{d}, J=2.4$ $\mathrm{Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 28.6,29.2,30.0,40.8,45.2,56.3,62.2,80.2$, 128.9, 131.1, 131.5, 135.9, 138.4, 141.0, 147.9, 155.2.

3'-(4-Chlorophenyl)epibatidine (5d) Hydrochloride. Compound 7b ( $247 \mathrm{mg}, 0.59 \mathrm{mmol}$ ) was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{~mL})$. TFA (3 mL ) was added dropwise at $0^{\circ} \mathrm{C}$ over 30 min . The mixture was stirred at room temperature for 3 h , poured into a cold solution of $\mathrm{NH}_{4} \mathrm{OH}-\mathrm{H}_{2} \mathrm{O}$ (1:1), and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic phase was washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and evaporated to dryness. Flash chromatography of the resulting residue on a silica gel column using $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$ yielded $147 \mathrm{mg}(78 \%)$ of $\mathbf{5 d}$ as a yellow oil: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.5-1.7(5 \mathrm{H}, \mathrm{m}), 1.94(1 \mathrm{H}, \mathrm{dd}, J=9.0,12.0 \mathrm{~Hz}), 2.82$ $(1 \mathrm{H}, \mathrm{dd}, J=5.1,9.0 \mathrm{~Hz}), 3.62(1 \mathrm{H}, \mathrm{brs}), 3.80(1 \mathrm{H}, \mathrm{brs}), 7.3-7.5(4 \mathrm{H}$, m), $7.76(1 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz}), 8.30(1 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 30.4,31.7,40.6,44.8,56.8,63.1,128.6,128.8,129.0,131.1$, 131.2, 131.5, 138.9, 141.8, 148.1.

Compound 5d ( $147 \mathrm{mg}, 0.46 \mathrm{mmol}$ ) was dissolved in MeOH ( 4.6 $\mathrm{mL})$ at room temperature. $\mathrm{HCl}(1 \mathrm{M}$ in ether, 4.6 mL$)$ was added with a syringe pump over 50 min at room temperature. After stirring for 30 min, the solvent was removed. The residue was recrystallized from a MeOH -ether mixture to give $\mathbf{5 d} \cdot \mathrm{HCl}$ as a yellow solid: $\mathrm{mp}>200^{\circ} \mathrm{C}$ (dec); anal. C $53.73 \%, \mathrm{H} 4.90 \%$, $\mathrm{N} 6.81 \%$, calcd for $\mathrm{C}_{17} \mathrm{H}_{17} \mathrm{Cl}_{3} \mathrm{~N}_{2} \cdot 1.5 \mathrm{H}_{2} \mathrm{O}$, C 53.35\%, H 5.27\%, N $7.32 \%$.

7-tert-Butoxycarbonyl-2-exo-[3-(4-nitrophenyl)-5-pyridinyl]-7azabicyclo[2.2.1]heptane (7c). To a resealable reaction tube were added compound 6 ( $233 \mathrm{mg}, 0.6 \mathrm{mmol}$ ), 4-nitrophenylboronic acid ( 200 mg , $1.2 \mathrm{mmol}), \mathrm{Pd}(\mathrm{OAc})_{2}(14 \mathrm{mg}, 0.06 \mathrm{mmol})$, tris $(o$-tolyl)phosphine ( 37 $\mathrm{mg}, 0.12 \mathrm{mmol}$ ), and $\mathrm{Na}_{2} \mathrm{CO}_{3}(159 \mathrm{mg}, 1.5 \mathrm{mmol})$ in DME ( 2 mL ) and $\mathrm{H}_{2} \mathrm{O}(0.5 \mathrm{~mL})$. The mixture was purged with argon, sealed, and heated in an $85^{\circ} \mathrm{C}$ oil bath overnight. The reaction mixture was cooled and diluted with EtOAc. The organic phase was washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated. Flash chromatography of the resulting residue on a silica gel column using EtOAc-hexanes (1:1) yielded $239 \mathrm{mg}(93 \%)$ of $\mathbf{7 c}$ as a yellowish oil: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.40(9 \mathrm{H}$, s), $1.5-1.9(5 \mathrm{H}, \mathrm{m}), 2.06(1 \mathrm{H}, \mathrm{m}), 2.97(1 \mathrm{H}, \mathrm{dd}, J=4.5,8.7 \mathrm{~Hz})$, $4.24(1 \mathrm{H}, \mathrm{brs}), 4.40(1 \mathrm{H}, \mathrm{brs}), 7.6-7.7(2 \mathrm{H}, \mathrm{m}), 7.70(1 \mathrm{H}, \mathrm{d}, J=2.4$ $\mathrm{Hz}), 8.25-8.35(2 \mathrm{H}, \mathrm{m}), 8.35(1 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ $\delta 28.6,29.2,30.0,40.8,45.1,56.3,62.2,80.3,123.8,124.2,130.8$, 131.1, 138.3, 141.3, 144.4, 149.4, 155.2.
$3^{\prime}$-(4-Nitrophenyl)epibatidine ( 5 g ) Hydrochloride. Compound 7c ( $239 \mathrm{mg}, 0.56 \mathrm{mmol}$ ) was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{~mL})$. TFA ( 3 mL ) was added dropwise at $0{ }^{\circ} \mathrm{C}$ over 30 min . The mixture was stirred at room temperature for 4 h , poured into a cold solution of $\mathrm{NH}_{4} \mathrm{OH}-\mathrm{H}_{2} \mathrm{O}$ (1:1), and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic phase was washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and evaporated to dryness. Flash chromatography of the resulting residue on a silica gel column using $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$ yielded $126 \mathrm{mg}(69 \%)$ of $\mathbf{5 g}$ as a yellow oil: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta$ $1.5-1.7(5 \mathrm{H}, \mathrm{m}), 1.97(1 \mathrm{H}, \mathrm{dd}, J=9.0,12.0 \mathrm{~Hz}), 2.72(1 \mathrm{H}, \mathrm{brs}), 2.85$ $(1 \mathrm{H}, \mathrm{dd}, J=5.1,9.0 \mathrm{~Hz}), 3.66(1 \mathrm{H}, \mathrm{brs}), 3.84(1 \mathrm{H}, \mathrm{m}), 7.6-7.7(2 \mathrm{H}$, $\mathrm{m}), 7.86(1 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz}), 7.25-7.35(2 \mathrm{H}, \mathrm{m}), 8.37(1 \mathrm{H}, \mathrm{d}, J=2.4$ $\mathrm{Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 30.4,31.7,40.6,44.7,56.9,63.2,123.9$, 130.9, 134.7, 138.8, 142.0, 144.6, 149.1.

Compound $\mathbf{5 g}(126 \mathrm{mg}, 0.38 \mathrm{mmol})$ was dissolved in $\mathrm{MeOH}(4 \mathrm{~mL})$ at room temperature. $\mathrm{HCl}(1 \mathrm{M}$ in ether, 4 mL$)$ was added with a syringe pump over 50 min at room temperature. After stirring for 30 min , the solvent was removed. The residue was recrystallized from $\mathrm{MeOH}-$ ether to give $\mathbf{5 g} \cdot \mathrm{HCl}$ as a yellow solid; mp $168-169^{\circ} \mathrm{C}$; anal. C $54.95 \%$, H $4.78 \%$, $\mathrm{N} 11.14 \%$, calcd for $\mathrm{C}_{17} \mathrm{H}_{17} \mathrm{Cl}_{2} \mathrm{~N}_{3} \mathrm{O}_{2} \cdot 0.25 \mathrm{H}_{2} \mathrm{O}, \mathrm{C} 55.07 \%, \mathrm{H}$ 4.76\%, N $11.33 \%$.

7-tert-Butoxycarbonyl-2-exo-[3-(4-tert-butoxylcarbonylaminophe-nyl)-5-pyridinyl]-7-azabicyclo[2.2.1]heptane (7d). To a resealable reaction tube were added compound $\mathbf{6}(231 \mathrm{mg}, 0.6 \mathrm{mmol}), 4-(N$-bocamino)phenylboronic acid ( $284 \mathrm{mg}, 1.2 \mathrm{mmol}$ ), $\mathrm{Pd}(\mathrm{OAc})_{2}(14 \mathrm{mg}, 0.06$ mmol ), tris ( $o$-tolyl)phosphine ( $37 \mathrm{mg}, 0.12 \mathrm{mmol}$ ), and $\mathrm{Na}_{2} \mathrm{CO}_{3}$ ( 159 $\mathrm{mg}, 1.5 \mathrm{mmol})$ in DME ( 2 mL ) and $\mathrm{H}_{2} \mathrm{O}(0.5 \mathrm{~mL})$. The mixture was purged with argon, sealed, and heated in an $85^{\circ} \mathrm{C}$ oil bath overnight. The reaction mixture was cooled, filtered through Celite, and diluted with EtOAc. The organic phase was washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated. Flash chromatography of the resulting residue on a silica gel column using EtOAc-hexanes (1:1) yielded 285 mg ( $96 \%$ ) of 7d: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.40(9 \mathrm{H}, \mathrm{s}), 1.53(9 \mathrm{H}, \mathrm{s}), 1.5-1.9(5 \mathrm{H}$,
m), $2.06(1 \mathrm{H}, \mathrm{m}), 2.90(1 \mathrm{H}, \mathrm{dd}, J=4.8,9.0 \mathrm{~Hz}), 4.22(1 \mathrm{H}, \mathrm{brs}), 4.38$ $(1 \mathrm{H}$, brs $), 6.78(1 \mathrm{H}, \mathrm{brs}), 7.3-7.5(4 \mathrm{H}, \mathrm{m}), 7.62(1 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz})$, $8.24(1 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 28.6,28.7,29.1,30.1$, $40.7,45.3,56.3,62.2,80.2,81.1,118.4,130.4,132.4,136.4,138.5$, 138.9, 140.8, 147.3, 147.8, 153.0, 155.3.
$\mathbf{3}^{\mathbf{\prime}}$-(4-Aminophenyl)epibatidine (5i) Dihydrochloride. Compound 7d ( $285 \mathrm{mg}, 0.57 \mathrm{mmol}$ ) was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{~mL})$. TFA ( 3 mL ) was added at $0^{\circ} \mathrm{C}$ over 30 min . The mixture was stirred at room temperature for 4 h , poured into a cold solution of concentrated $\mathrm{NH}_{4} \mathrm{OH}-\mathrm{H}_{2} \mathrm{O}(1: 1)$, and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic phase was washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and evaporated to dryness. Flash chromatography of the resulting residue on a silica gel column using $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$ yielded $143 \mathrm{mg}(84 \%)$ of $\mathbf{5 i}$ as a yellow oil: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.5-1.7(5 \mathrm{H}, \mathrm{m}), 1.91(1 \mathrm{H}, \mathrm{dd}, J=9.0,12.0 \mathrm{~Hz}), 2.80$ $(1 \mathrm{H}, \mathrm{dd}, J=5.1,9.0 \mathrm{~Hz}), 3.60(1 \mathrm{H}, \mathrm{m}), 3.77(1 \mathrm{H}, \mathrm{m}), 6.6-6.8(2 \mathrm{H}$, $\mathrm{m}), 7.2-7.3(2 \mathrm{H}, \mathrm{m}), 7.70(1 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz}), 8.22(1 \mathrm{H}, \mathrm{d}, J=2.4$ $\mathrm{Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 30.4,31.6,40.6,45.0,56.8,63.1,114.9$, $127.8,130.8,136.7,13838,141.6,146.9,147.0,147.5$.

The free base $\mathbf{5 i}$ ( $143 \mathrm{mg}, 0.48 \mathrm{mmol}$ ) was dissolved in $\mathrm{MeOH}(5$ $\mathrm{mL})$ at room temperature. $\mathrm{HCl}(1 \mathrm{M}$ in ether, 5 mL$)$ was added with a syringe pump over 50 min at room temperature. After stirring for 30 min, the solvent was removed, and the residue was recrystallized from $\mathrm{MeOH}-$ ether to give $\mathbf{5 i} \cdot \mathrm{HCl}$ as a yellow solid: $\mathrm{mp}>265^{\circ} \mathrm{C}$ (dec); anal. C $50.89 \%$, H $5.86 \%, \mathrm{~N} 10.14 \%$, calcd for $\mathrm{C}_{17} \mathrm{H}_{20} \mathrm{Cl}_{3} \mathrm{~N}_{3} \cdot 1.5 \mathrm{H}_{2} \mathrm{O}$, C $51.08 \%$, H $5.80 \%$, N $10.51 \%$.

7-tert-Butoxycarbonyl-2-exo-[3-(4-methoxyphenyl)-5-pyridinyl]-7-azabicyclo[2.2.1]heptane (7e). To a resealable reaction tube were added compound $\mathbf{6}(231 \mathrm{mg}, 0.6 \mathrm{mmol})$, 4-methoxyphenylboronic acid $(182 \mathrm{mg}, 1.2 \mathrm{mmol}), \mathrm{Pd}(\mathrm{OAc})_{2}(14 \mathrm{mg}, 0.06 \mathrm{mmol})$, tris $(o$-tolyl)phosphine ( $37 \mathrm{mg}, 0.12 \mathrm{mmol}$ ), and $\mathrm{Na}_{2} \mathrm{CO}_{3}(159 \mathrm{mg}, 1.5 \mathrm{mmol})$ in DME $(2 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}(0.5 \mathrm{~mL})$. The mixture was purged with argon, sealed, and heated in an $85^{\circ} \mathrm{C}$ oil bath overnight. The reaction mixture was cooled, filtered through Celite, and diluted with EtOAc. The organic phase was washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated. Flash chromatography of the resulting residue on a silica gel column using EtOAc-hexanes (1:1) yielded $223 \mathrm{mg}(90 \%)$ of $7 \mathbf{e}$ as a yellow oil: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.40(9 \mathrm{H}, \mathrm{s}), 1.5-1.9(5 \mathrm{H}, \mathrm{m}), 2.01(1 \mathrm{H}, \mathrm{dd}, J=9.0$, 12.3 Hz ), $2.91(1 \mathrm{H}, \mathrm{dd}, J=4.8,8.7 \mathrm{~Hz}), 3.84(3 \mathrm{H}, \mathrm{s}), 4.22(1 \mathrm{H}, \mathrm{brs})$, $4.38(1 \mathrm{H}, \mathrm{brs}), 6.9-7.0(2 \mathrm{H}, \mathrm{m}), 7.3-7.4(2 \mathrm{H}, \mathrm{m}), 7.64(1 \mathrm{H}, \mathrm{d}, J=$ $2.4 \mathrm{~Hz}), 8.23(1 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 28.6,29.2$, 30.1, 40.7, 45.2, 55.7, 56.3, 62.2, 80.1, 114.1, 130.2, 130.9, 136.5, 138.6, 140.8, 147.2, 155.2, 160.0.
$\mathbf{3}^{\prime}$-(4-Methoxylphenyl)epibatidine ( 5 k ) Dihydrochloride. Compound $7 \mathbf{e}(223 \mathrm{mg}, 0.54 \mathrm{mmol})$ was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{~mL})$. TFA $(3 \mathrm{~mL})$ was added dropwise at $0{ }^{\circ} \mathrm{C}$ over 30 min . The mixture was stirred at room temperature for 4 h , poured into a cold solution of concentrated $\mathrm{NH}_{4} \mathrm{OH}-\mathrm{H}_{2} \mathrm{O}$ (1:1), and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic phase was washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and evaporated to dryness. Flash chromatography of the resulting residue on a silica gel column using $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$ yielded $128 \mathrm{mg}(76 \%)$ of $\mathbf{5 k}$ as a yellow oil: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.5-1.7(5 \mathrm{H}, \mathrm{m}), 1.92(1 \mathrm{H}, \mathrm{dd}, J=$ $9.0,12.3 \mathrm{~Hz}), 2.80(1 \mathrm{H}, \mathrm{dd}, J=4.8,8.7 \mathrm{~Hz}), 3.60(1 \mathrm{H}, \mathrm{brs}), 3.78(1 \mathrm{H}$, $\mathrm{t}, J=3.6 \mathrm{~Hz}), 3.84(3 \mathrm{H}, \mathrm{s}), 6.9-7.0(2 \mathrm{H}, \mathrm{m}), 7.3-7.4(2 \mathrm{H}, \mathrm{m}), 7.74$ $(1 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz}), 8.26(1 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta$ $30.1,31.4,40.3,44.6,55.3,56.4,62.8,113.7,130.1,130.6,136.0,138.6$, $141.4,147.0,147.2,159.530 .2,31.7,40.6,44.8,56.8,63.1,115.6$ (d, $J=0.855 \mathrm{~Hz}), 131.6(\mathrm{~d}, J=0.33 \mathrm{~Hz}), 135.7,138.9,141.8,147.5$, 148.0, 161.4, 164.6.

The free base $\mathbf{5 k}$ ( $128 \mathrm{mg}, 0.41 \mathrm{mmol}$ ) was dissolved in $\mathrm{MeOH}(4$ $\mathrm{mL})$ at room temperature. $\mathrm{HCl}(1 \mathrm{M}$ in ether, 4 mL$)$ was added with a syringe pump over 20 min at room temperature. After stirring for 30 min , the solvent was removed, and the residue was recrystallized from a MeOH -ether mixture to give $\mathbf{5 k} \cdot \mathbf{2 H C l}$ as a yellow solid: $\mathrm{mp}>200$ ${ }^{\circ} \mathrm{C}$ (dec); anal. C $53.62 \%, \mathrm{H} 5.63 \%$, $\mathrm{N} 5.90 \%$, calcd for $\mathrm{C}_{18} \mathrm{H}_{21} \mathrm{Cl}_{3} \mathrm{~N}_{2} \mathrm{O} \cdot \mathrm{H}_{2} \mathrm{O}$, C $53.28 \%$, H $5.71 \%$, $\mathrm{N} 6.90 \%$.
3'-(3-Aminophenyl)epibatidine (5j) Dihydrochloride. Compound 5h ( $80 \mathrm{mg}, 0.241 \mathrm{mmol}$ ), ethanol ( 2 mL ), water ( 0.06 mL ), and concentrated $\mathrm{HCI}(0.01 \mathrm{~mL})$ were stirred at room temperature for 10 min . Iron powder ( $149.2 \mathrm{mg}, 2.66 \mathrm{mmol}$ ) was added in small portions. The mixture was heated at $100{ }^{\circ} \mathrm{C}$ for 30 min , poured into a cold solution of sodium carbonate (aqueous solution), and extracted with ethyl acetate. The organic phase was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The residue was purified by silica gel column chromatography eluting with CMA80-EtOAc (1:3) to yield $70 \mathrm{mg}(95 \%)$ of $\mathbf{5 j}$ as a yellow
oil: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.48-1.70(5 \mathrm{H}, \mathrm{m}), 1.87-1.99(1 \mathrm{H}, \mathrm{m})$, $2.79-2.84(1 \mathrm{H}, \mathrm{dd}), 3.61(1 \mathrm{H}, \mathrm{brs}), 3.78(1 \mathrm{H}$, brs $), 6.70-6.74(2 \mathrm{H}$, m), $6.81(1 \mathrm{H}, \mathrm{d}, J=6.0 \mathrm{~Hz}), 7.20(1 \mathrm{H}, \mathrm{d}, J=6.0 \mathrm{~Hz}), 7.72(\mathrm{~s}, 11: 1)$, $8.27(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 30.3,31.6,40.6,44.9,56.8,63.1$, $115.2,116.3,119.9,129.5,137.0,138.9,139.2,141.5,146.7,147.4$, 147.7.

Compound $\mathbf{5 j}$ ( $70 \mathrm{mg}, 0.231 \mathrm{mmol}$ ) was dissolved in 4 mL of methanol, and 1 M HCI in ether ( 2 mL ) was added. After stirring for 30 min , the solvent was removed. The residue was recrystallized from MeOH -ether ( $1: 3$ ) to yield $\mathbf{5 j} \cdot \mathrm{HCl}$ as a yellow solid: $\mathrm{mp} 195^{\circ} \mathrm{C}$ (dec); anal. C $51.08 \%$, H $5.80 \%, \mathrm{~N} 10.51 \%$, calcd for $\mathrm{C}_{17} \mathrm{H}_{20} \mathrm{Cl}_{3} \mathrm{~N}_{3} \cdot 1.5 \mathrm{H}_{2} \mathrm{O}$, C $51.14 \%$, H $5.91 \%$, N $9.61 \%$.

7-tert-Butoxycarbonyl-2-exo-[3-(3-nitrophenyl)-5-pyridinyl]-7azabicyclo[2.2.1]heptane (8). A solution of compound $\mathbf{5 h}(0.31 \mathrm{~g}, 0.94$ mmol ), tert-butoxycarbonyl anhydride ( $400 \mathrm{mg}, 1.83 \mathrm{mmol}$ ), DMAP $(10 \mathrm{mg})$, triethylamine $(0.1 \mathrm{~mL})$, and methylene chloride ( 5 mL ) was stirred for 1 h , poured into 100 mL of $1 \mathrm{M} \mathrm{K}_{2} \mathrm{CO}_{3}$, and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic phase was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The residue was purified by flash chromatography on a silica gel column eluting with hexane-EtOAc (3:1) to yield $0.25 \mathrm{~g}(60 \%)$ of $\mathbf{8}$ as an oil: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.39(9 \mathrm{H}, \mathrm{s}), 1.50-1.70(5 \mathrm{H}, \mathrm{m}), 1.96-2.09(1 \mathrm{H}$, $\mathrm{m}), 2.93-2.97(1 \mathrm{H}, \mathrm{m}),, 4.23(1 \mathrm{H}$, brs $), 4.39(1 \mathrm{H}$, brs), $7.64-7.70(2 \mathrm{H}$, $\mathrm{m}), 7.79-7.80(1 \mathrm{H}, \mathrm{m}), 8.31-8.35(1 \mathrm{H}, \mathrm{m}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 27.8$, $28.6\left(\mathrm{CH}_{3}-3\right), 31.3,40.8,45.2,56.4,62.3,80.4,123.6,124.7,129.8$, $134.5,135.9,138.5,139.6,141.4,147.5,148.8,149.1,155.4$.
7-tert-Butoxycarbonyl-2-exo-[3-(3-aminophenyl)-5-pyridinyl]-7azabicyclo[2.2.1]heptane (9). To a mixture of compound 8 ( 80 mg , $0.81 \mathrm{mmol})$ and $\mathrm{Ni}_{2} \mathrm{~B}\left[56.1 \mathrm{mg}\right.$, prepared from $\left.\mathrm{Ni}(\mathrm{OAc})_{2}\right]$ in MeOH $(3.2 \mathrm{~mL})$ was added $1 \mathrm{M} \mathrm{HCl}(0.8 \mathrm{~mL})$. The reaction mixture was heated at $60^{\circ} \mathrm{C}$ for 30 min , poured into 100 mL of a solution of concentrated $\mathrm{NH}_{4} \mathrm{OH}-\mathrm{H}_{2} \mathrm{O}$ (1:1), extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated. The residue was purified by flash chromatography eluting with hexane-EtOAc (3:1) to yield 70 mg (93\%) of 9 as a yellow oil: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDC1}_{3}\right) \delta 1.40(9 \mathrm{H}, \mathrm{s}), 1.51-1.85(5 \mathrm{H}$, $\mathrm{m}), 1.97-2.04(1 \mathrm{H}, \mathrm{m}), 2.87-2.92(1 \mathrm{H}, \mathrm{m}), 3.77(2 \mathrm{H}, \mathrm{brs}), 4.21(1 \mathrm{H}$, brs), $4.38(1 \mathrm{H}$, brs $), 6.70-6.74(2 \mathrm{H}, \mathrm{m}), 6.78-6.81(1 \mathrm{H}, \mathrm{m}), 7.20(1 \mathrm{H}$, $\mathrm{t}, J=6.0 \mathrm{~Hz}), 7.62(1 \mathrm{H}, \mathrm{s}), 8.25(1 \mathrm{H}, \mathrm{s}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 28.7$ $\left(\mathrm{CH}_{3}-3\right), 29.2,30.2,40.7,45.3,56.3,62.3,80.3,115.3,116.3,120.0$, 129.6, 137.1, 138.6, 139.1, 140.7, 146.7, 147.5, 147.8.

7-tert-Butoxycarbonyl-2-exo-[3'-(3-dimethylamino)-5-pyridinyl]-7-azabicyclo[2.2.1]heptane (10). A mixture of 9 ( $180 \mathrm{mg}, 0.433 \mathrm{mmol}$ ), acetonitrile ( 12 mL ), $37 \%$ aqueous formaldehyde ( 1.54 mL ), and $\mathrm{NaCNBH}_{3}(488 \mathrm{mg}, 7.76 \mathrm{mmol})$ was stirred for 3 h at room temperature. Glacial acetic acid ( 0.642 mL ) was added and stirring continued overnight. The reaction mixture was poured into 100 mL of a solution of concentrated $\mathrm{NH}_{4} \mathrm{OH}-\mathrm{H}_{2} \mathrm{O}(1: 1)$, extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated. The residue was purified by flash chromatography on a silica gel column eluting with hexanes-EtOAc (3:1) to yield $179 \mathrm{mg}(93 \%)$ of $\mathbf{1 0}$ as a yellow oil: ${ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ $\delta 1.39(9 \mathrm{H}, \mathrm{s}), 1.54-1.89(5 \mathrm{H}, \mathrm{m}), 1.97-2.01(1 \mathrm{H}, \mathrm{m}), 2.88-2.93$ $(1 \mathrm{H}, \mathrm{m}), 2.98(6 \mathrm{H}, \mathrm{s}), 4.22(1 \mathrm{H}, \mathrm{brs}), 4.36(1 \mathrm{H}$, brs $), 6.73-6.79(3 \mathrm{H}$, m), $7.29(1 \mathrm{H}, \mathrm{d}, J=6.0 \mathrm{~Hz}), 7.65(1 \mathrm{H}, \mathrm{s}), 8.26(1 \mathrm{H}, \mathrm{s})$.

3'-(3-Dimethylaminophenyl)epibatidine (5m) Dihydrochloride. Compound 10 ( $179 \mathrm{mg}, 0.403 \mathrm{mmol}$ ) in methylene chloride ( 3 mL ) was stirred at $0^{\circ} \mathrm{C}$ for 15 min ; then trifluoroacetic acid ( 3 mL ) was added. After stirring for 30 min , the reaction mixture was poured into 100 mL of a solution of concentrated $\mathrm{NH}_{4} \mathrm{OH}-\mathrm{H}_{2} \mathrm{O}(1: 1)$, extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated. The residue was purified by flash chromatography on a silica gel column eluting with CMA-EtOAc (1:5) to yield $132 \mathrm{mg}(95 \%)$ of $\mathbf{5 m}$ : ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ $\delta 1.49-1.69(5 \mathrm{H}, \mathrm{m}), 1.89-1.96(1 \mathrm{H}, \mathrm{m}), 2.79-2.84(1 \mathrm{H}, \mathrm{m}), 2.98$ ( $6 \mathrm{H}, \mathrm{s}$ ), $3.61(1 \mathrm{H}, \mathrm{brs}), 3.77(1 \mathrm{H}, \mathrm{brs}), 6.75-6.78(3 \mathrm{H}, \mathrm{m}), 7.3(1 \mathrm{H}, \mathrm{d}$, $J=6.0 \mathrm{~Hz}), 7.75(1 \mathrm{H}, \mathrm{s}), 8.29(1 \mathrm{H}, \mathrm{s}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 30.5$, 31.7, 40.7, $41.0\left(2 \mathrm{CH}_{3}\right), 45.1,56.8,63.1,112.6,113.9,117.9,129.4$, 137.6, 139.0, 141.6, 147.5, 147.6, 150.7. (One tertiary aromatic C was not observed.)

Compound $\mathbf{5 m}$ ( $132 \mathrm{mg}, 0.383 \mathrm{mmol}$ ) was dissolved in 4 mL of methanol, and 1 M HCl in ether ( 4 mL ) was added dropwise. After concentration the residue was recrystallized from a MeOH -ether mixture (1:3) to give $\mathbf{5 m} \cdot 2 \mathrm{HCl}$ : mp $69-72{ }^{\circ} \mathrm{C}$; anal. $\mathrm{C} 54.23 \%$, H $6.49 \%$, $\mathrm{N} 9.51 \%$, calcd for $\mathrm{C}_{19} \mathrm{H}_{24} \mathrm{Cl}_{3} \mathrm{~N}_{3} \cdot 1.25 \mathrm{H}_{2} \mathrm{O}, \mathrm{C} 53.91 \%$, $\mathrm{H} 6.31 \%$, N $9.93 \%$.

3'-(3-Fluorophenyl)epibatidine (5c) Hydrochloride. To a solution of $\mathbf{5 j}(150 \mathrm{mg}, 0.5 \mathrm{mmol})$ in $70 \% \mathrm{HF}-$ pyridine $(2.7 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was
added $\mathrm{NaNO}_{2}$ ( $266 \mathrm{mg}, 3.9 \mathrm{mmol}$ ). The reaction mixture was stirred for 30 min at $0{ }^{\circ} \mathrm{C}$ and then heated to $100^{\circ} \mathrm{C}$ for an additional hour. The reaction mixture was poured into 100 mL of a solution of $\mathrm{NH}_{4} \mathrm{OH}-\mathrm{H}_{2} \mathrm{O}$ (1:1), extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated. The residue was purified by flash chromatography on a silica gel column eluting with CMA-EtOAc (1:3) to yield 40 mg (26\%) of 5 c as an oil: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.53-1.67(5 \mathrm{H}, \mathrm{m}), 1.90-2.02$ $(1 \mathrm{H}, \mathrm{m}), 2.79-2.83(1 \mathrm{H}, \mathrm{m}), 3.61(1 \mathrm{H}, \mathrm{brs}), 3.80(1 \mathrm{H}, \mathrm{brs}), 7.08-7.24$ $(1 \mathrm{H}, \mathrm{m}), 7.38-7.42(1 \mathrm{H}, \mathrm{m}), 7.78(1 \mathrm{H}, \mathrm{s}), 8.31\left(1 \mathrm{H}, \mathrm{s}\right.$, pyridinyl); ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 30.6,31.9,40.8,44.9,56.8,63.2,115.4\left(\mathrm{~d}, J_{\mathrm{CF}}=\right.$ $20.9 \mathrm{~Hz}), 116.9\left(\mathrm{~d}, J_{\mathrm{CF}}=22.3 \mathrm{~Hz}\right), 125.6\left(\mathrm{~d}, J_{\mathrm{CF}}=2.9 \mathrm{~Hz}\right), 130.2(\mathrm{~d}$, $\left.J_{\mathrm{CF}}=8.2 \mathrm{~Hz}\right), 135.5,138.9,140.2\left(\mathrm{~d}, J_{\mathrm{CF}}=7.7 \mathrm{~Hz}\right), 142.1,147.3$, 148.3, 162.8 (d, $J_{\mathrm{CF}}=245 \mathrm{~Hz}$ ).

Compound 5c ( $40 \mathrm{mg}, 0.131 \mathrm{mmol}$ ) was dissolved in 4 mL of methanol, and 1 M HCl in ether ( 4 mL ) was added dropwise. The residue obtained on concentration was recrystallized from methanol and ether to give $\mathbf{5 c} \cdot \mathrm{HCl}$ : $\mathrm{mp} 101-105^{\circ} \mathrm{C}$; anal. C $57.53 \%, \mathrm{H} 5.43 \%$, N $7.66 \%$, calcd for $\mathrm{C}_{17} \mathrm{H}_{17} \mathrm{Cl}_{2} \mathrm{FN}_{2} \cdot 0.75 \mathrm{H}_{2} \mathrm{O}$, C $57.88 \%$, H $5.29 \%$, N 7.94\%.
$\mathbf{3}^{\mathbf{\prime}}$-(3-Chlorophenyl)epibatidine (5e) Hydrochloride. To a mixture of anhydrous $\mathrm{CuCl}(32.3 \mathrm{mg}, 0.24 \mathrm{mmol})$, butyl nitrite $(0.04 \mathrm{~mL}, 0.3$ $\mathrm{mmol})$, and anhydrous acetonitrile ( 10 mL ) warmed to $65^{\circ} \mathrm{C}$ was added $\mathbf{5 j}(60.4 \mathrm{mg}, 0.2 \mathrm{mmol})$ in 2 mL of anhydrous acetonitrile over 5 min . After 20 min , the cooled reaction mixture was poured into 10 mL of $20 \% \mathrm{HCl}(\mathrm{aq})$, stirred 10 min , poured into 100 mL of a solution of concentrated $\mathrm{NH}_{4} \mathrm{OH}-\mathrm{H}_{2} \mathrm{O}$ (1:1), extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated. The residue was purified by flash chromatography on a silica gel column eluting with CMA-EtOAc (1:5) to yield $50 \mathrm{mg}(78 \%)$ of $\mathbf{5 e}:{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.50-1.72(5 \mathrm{H}, \mathrm{m})$, $1.90-1.97(1 \mathrm{H}, \mathrm{m}), 2.79-2.83(1 \mathrm{H}, \mathrm{m}), 3.61(1 \mathrm{H}$, brs), $3.79(1 \mathrm{H}, \mathrm{brs})$, $7.34-7.43(4 \mathrm{H}, \mathrm{m}), 7.77(1 \mathrm{H}, \mathrm{s}), 8.30(1 \mathrm{H}, \mathrm{s}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta$ $30.6,31.8,40.8,44.9,56.8,63.2,128.1,128.6,129.8,129.9,134.5$, $135.4,138.9,139.9,142.1,147.3,148.4$.

Compound 5e ( $50 \mathrm{mg}, 0.383 \mathrm{mmol}$ ) was dissolved in 4 mL of methanol, and 1 M HCI in ether ( 2 mL ) was added. The residue obtained on concentration was recrystallized from $\mathrm{CH}_{3} \mathrm{OH}$-ether to give $\mathbf{5 e} \cdot \mathrm{HCl}$ : mp $159{ }^{\circ} \mathrm{C}$ (dec); anal. C $54.78 \%$, H 5.08\%, N $7.18 \%$, calcd for $\mathrm{C}_{17} \mathrm{H}_{17} \mathrm{Cl}_{3} \mathrm{~N}_{2} \cdot \mathrm{H}_{2} \mathrm{O}$ C $54.64 \%$, H $5.12 \%$, $\mathrm{N} 7.50 \%$.
$\mathbf{3}^{\prime}$-(3-Bromophenyl)epibatidine (5f) Hydrochloride. To a mixture of anhydrous $\mathrm{CuBr}(53.6 \mathrm{mg}, 0.24 \mathrm{mmol})$, butyl nitrite $(0.04 \mathrm{~mL}, 0.3$ mmol ), and anhydrous acetonitrile ( 6 mL ), warmed to $65^{\circ} \mathrm{C}$, was added $5 \mathbf{j}(60.4 \mathrm{mg}, 0.2 \mathrm{mmol})$ in 4 mL of anhydrous acetonitrile. After 20 min , the reaction mixture was poured into 10 mL of $20 \% \mathrm{HCl}(\mathrm{aq})$, stirred 10 min , poured into 100 mL of a solution of concentrated $\mathrm{NH}_{4} \mathrm{OH}-\mathrm{H}_{2} \mathrm{O}$ (1:1), extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated. The residue obtained was purified by flash chromatography on a silica gel column eluting with CMA-EtOAc (1:5) to yield $30 \mathrm{mg}(41 \%)$ of $\mathbf{5 f}:{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.50-1.73(5 \mathrm{H}, \mathrm{m}), 1.90-1.99$ $(1 \mathrm{H}, \mathrm{m}), 2.79-2.83(1 \mathrm{H}, \mathrm{m}), 3.61(1 \mathrm{H}, \mathrm{brs}), 3.80(1 \mathrm{H}, \mathrm{brs}), 7.31-7.41$ $(4 \mathrm{H}, \mathrm{m}), 7.77(1 \mathrm{H}, \mathrm{s}), 8.31(1 \mathrm{H}, \mathrm{s}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 30.6,31.9$, $40.8,44.9,56.8,63.2,122.6,128.5,130.2,131.6,132.6,134.5,135.3$, 138.9, 140.2, 142.1, 148.4.

Compound $\mathbf{5 f}$ ( $60 \mathrm{mg}, 0.164 \mathrm{mmol}$ ) was dissolved in 4 mL of methanol, and 1 M HCl in ether ( 3 mL ) was added. The residue obtained on concentration was recrystallized from $\mathrm{CH}_{3} \mathrm{OH}-$ ether to give $\mathbf{5 f} \cdot \mathrm{HCl}$ : $\mathrm{mp} 117-120^{\circ} \mathrm{C}$ (dec); anal. C $49.57 \%, \mathrm{H} 4.66 \%, \mathrm{~N}, 6.55 \%$, calcd for $\mathrm{C}_{17} \mathrm{H}_{17} \mathrm{BrCl}_{2} \mathrm{~N}_{2} \cdot 0.75 \mathrm{H}_{2} \mathrm{O}$, C $49.36 \%$, H $4.51 \%$, N $6.77 \%$.
$\left[{ }^{3} \mathbf{H}\right]$ Epibatidine Binding Assay. Adult male rat cerebral cortices (Pelfreeze Biological, Rogers, AK) were homogenized in 39 volumes of ice-cold 50 mM Tris buffer ( pH 7.4 at $4^{\circ} \mathrm{C}$ ) containing 120 mM $\mathrm{NaCl}, 5 \mathrm{mM} \mathrm{KCl}, 2 \mathrm{mM} \mathrm{CaCl} 2$, and 1 mM MgCl 2 and sedimented at 37000 g for 10 min at $4^{\circ} \mathrm{C}$. The supernatant was discarded, the pellet resuspended in the original volume of buffer, and the wash procedure repeated twice more. After the last centrifugation, the pellet was resuspended in one-tenth its original homogenization volume and stored at $-80^{\circ} \mathrm{C}$ until needed. In a final volume of 0.5 mL , each assay tube contained 3 mg wet weight male rat cerebral cortex homogenate (added last), $0.5 \mathrm{nM}\left[{ }^{3} \mathrm{H}\right]$ epibatidine (NEN Life Science Products, Wilmington, DE), and one of 10-12 different concentrations of test compound dissolved in buffer ( pH 7.4 at room temperature) containing $10 \%$ DMSO, resulting in a final DMSO concentration of $1 \%$. Total and nonspecific bindings were determined in the presence of vehicle and $300 \mu \mathrm{M}(-)$-nicotine, respectively. After a 4 h incubation period at room temperature, the samples were vacuum-filtered over GF/B filter papers presoaked in $0.03 \%$ polyethylenimine using a Brandel 48 -well
harvester and washed with 6 mL of ice-cold buffer. The amount of radioactivity trapped on the filter was determined by standard liquid scintillation techniques in a TriCarb 2200 scintillation counter (Packard Instruments, Meriden, CT) at approximately $50 \%$ efficiency. The binding data were fit using the nonlinear regression analysis routines in Prism (Graphpad, San Diego, CA). The $K_{\mathrm{i}}$ values for the test compounds were calculated from their respective $\mathrm{IC}_{50}$ values using the Cheng-Prusoff equation.
[ ${ }^{125}$ I]Iodo-MLA Binding Assay. Adult male rat cerebral cortices (Pel-Freez Biologicals, Rogers, AK) were homogenized (polytron) in 39 volumes of ice-cold 50 mM Tris buffer (assay buffer; pH 7.4 at 4 ${ }^{\circ} \mathrm{C}$ ) containing $120 \mathrm{mM} \mathrm{NaCl}, 5 \mathrm{mM} \mathrm{KCl}, 2 \mathrm{mM} \mathrm{CaCl} 2$, and 1 mM $\mathrm{MgCl}_{2}$. The homogenate was centrifuged at 35000 g for 10 min at $4{ }^{\circ} \mathrm{C}$ and the supernatant discarded. The pellet was resuspended in the original volume of buffer and the wash procedure repeated twice more. After the last centrifugation step, the pellet was resuspended in one-tenth the original homogenization volume and stored at $-80^{\circ} \mathrm{C}$ until needed. Triplicate samples were run in 1.4 mL polypropylene tubes (Matrix Technologies Corporation, Hudson, NH). Briefly, in a final volume of 0.5 mL , each assay sample contained 3 mg wet weight rat cerebral cortex (added last), $40-50 \mathrm{pM}\left[{ }^{125} \mathrm{I}\right]$ MLA, and 50 nM final concentration of test compound dissolved in buffer containing $10 \%$ DMSO, giving a final DMSO concentration of $1 \%$. Total and nonspecific binding were determined in the presence of vehicle and $300 \mu \mathrm{M}(-)-$ nicotine, respectively. After a 2 h incubation period on ice, the samples were vacuum-filtered using a Multimate 96 -well harvester (Packard Instruments, Meriden, CT) onto GF/B filters presoaked for at least 30 min in assay buffer containing $0.15 \%$ bovine serum albumin. Each well was then washed with approximately 3.0 mL of ice-cold buffer. The filter plates were dried, and $30 \mu \mathrm{~L}$ of Microscint20 (Packard) was added to each well. The amount of radioligand remaining on each filter was determined using a TopCount 12 -detector (Packard) microplate scintillation counter at approximately $70 \%$ efficiency.

Tail-Flick Test. Antinociception was assessed by the tail-flick method of D'Amour and Smith. ${ }^{21}$ A control response ( $2-4$ s) was determined for each mouse before treatment, and a test latency was determined after drug administration. In order to minimize tissue damage, a maximum latency of 10 s was imposed. Antinociceptive response was calculated as percent maximum possible effect (\% MPE), where $\%$ MPE $=[($ test - control $) /(10-$ control $)] \times 100$. Groups of eight to 12 animals were used for each dose and for each treatment. The mice were tested 5 min after sc injections of epibatidine analogues for the dose-response evaluation. Eight to 12 mice were treated per dose, and a minimum of four doses were performed for dose-response curve determination.

Hot-Plate Test. Mice were placed into a 10 cm wide glass cylinder on a hot plate (Thermojust Apparatus) maintained at $55.0^{\circ} \mathrm{C}$. Two control latencies at least 10 min apart were determined for each mouse. The normal latency (reaction time) was 8 to 12 s . Antinociceptive response was calculated as percent maximum possible effect (\% MPE), where $\%$ MPE $=[($ test - control $) / 40-$ control $) \times 100]$. The reaction time was scored when the animal jumped or licked its paws. Eight mice per dose were injected sc with epibatidine analogues and tested 5 min thereafter in order to establish a dose-response curve.

Locomotor Activity. Mice were placed into individual Omnitech photocell activity cages ( $28 \times 16.5 \mathrm{~cm}$ ) 5 min after sc administration of either $0.9 \%$ saline or epibatidine analogues. Interruptions of the photocell beams (two banks of eight cells each) were then recorded for the next 10 min . Data were expressed as number of photocell interruptions.

Body Temperature. Rectal temperature was measured by a thermistor probe (inserted 24 mm ) and digital thermometer (Yellow Springs Instrument Co., Yellow Springs, OH). Readings were taken just before and 30 min after the sc injection of either saline or epibatidine analogues. The difference in rectal temperature before and after treatment was calculated for each mouse. The ambient temperature of the laboratory varied from 21 to $24^{\circ} \mathrm{C}$ from day to day.

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